404 INFLUENCE OF HYDROSTATIC PRESSURE ON THE STRUCTURE OF BaFCI

References

- BARNETT, J. D., BLOCK, S. & PIERMARINI, G. J. (1973). Rev. Sci. Instrum. 44, 1–9.
- BÄRNIGHAUSEN, H., BRAUER, G. & SCHULTZ, N. (1965). Z. Anorg. Allg. Chem. 338, 250–265.
- BECK, H. P. (1976). J. Solid State Chem. 17, 275-282.
- BECK, H. P. (1978a). Habilitationsschrift, Karlsruhe.
- BECK, H. P. (1978b). J. Solid State Chem. 23, 213-217.
- BECK, H. P. (1979). Z. Anorg. Allg. Chem. 451, 73-81.
- CROMER, D. T. & MANN, J. B. (1968). Acta Cryst. A24, 321-324.
- DECKER, D. L. (1971). J. Appl. Phys. 42, 3239-3244.
- DENNER, W., DIETERICH, W., SCHULZ, H., KELLER, R. & HOLZAPFEL, W. B. (1978). Rev. Sci. Instrum. 49, 775-777.
- DENNER, W., SCHULZ, H. & D'AMOUR, H. (1978). J. Appl. Cryst. 11, 260–264.
- FLAHAUT, J. (1974). J. Solid State Chem. 9, 124-131.

- GUTMANN, V. & MAYER, H. (1976). Struct. Bonding (Berlin), 31, 49-66.
- JOHNSON, V. & JEITSCHKO, W. (1973). J. Solid State Chem. 6, 306-309.
- KELLER, R. & HOLZAPFEL, W. B. (1977). Rev. Sci. Instrum. 48, 517–523.
- PIERMARINI, G. J., BLOCK, S. & BARNETT, J. D. (1973). J. Appl. Phys. 44, 5377-5382.
- SAUVAGE, M. (1974). Acta Cryst. B30, 2786-2787.
- STEWART, J. M., KUNDELL, F. A. & BALDWIN, J. C. (1970). The XRAY system. Computer Science Center, Univ. of Maryland, College Park, Maryland.
- TESSMANN, J. K., KAHN, A. H. & SHOCKLEY, W. (1953). Phys. Rev. 92, 890–895.
- ZUCKER, U. H., PERENTHALER, E., KUHS, W. F., BACHMANN, R. & SCHULZ, H. (1983). J. Appl. Cryst. 16, 358.

Acta Cryst. (1983). B39, 404–407

Anisotropic Mean-Square Displacements (MSD) in Single Crystals of 2H- and 3R-MoS₂

BY B. SCHÖNFELD,* J. J. HUANG[†] AND S. C. MOSS

University of Houston, Houston, Texas 77004, USA

(Received 25 October 1982; accepted 2 March 1983)

Abstract

Integrated intensities for h0.l reflections in single crystals of 2H- and 3R-MoS₂ have been measured both to refine the *c*-axis structural parameter (*z*) and to determine the vibrational amplitude for the Mo and S atoms. It was found that the MSD for the Mo atom along the *c* axis is distinctly larger than for the S atom $(7.6 \times 10^{-3} \text{ Å}^2 \text{ versus } 5.0 \times 10^{-3} \text{ Å}^2)$, independent of the polytype studied. Within the basal plane the amplitudes of 2H-MoS₂ are both smaller and more equal $(4.8 \times 10^{-3} \text{ Å}^2 \text{ for Mo and } 3.8 \times 10^{-3} \text{ Å}^2 \text{ for S})$. Both polytypes gave a *z* parameter for the S atom of $z = 0.127_4$.

I. Introduction

 MoS_2 belongs to the layered dichalcogenides of transition metals of Groups IVb, Vb and VIb, which have attracted interest on account of their highly anisotropic properties. A review of these materials may be found in Wilson & Yoffe (1969). The metal dichalcogenide layer consists of one plane of hexagonally packed metal atoms between two planes of

chalcogenide atoms. In these layers the chalcogen array around each metal atom is typically octahedral or trigonal prismatic and MoS_2 belongs to the latter group. The metal dichalcogenide layers may be stacked in different ways (polytypism). Thus, MoS_2 , either natural or synthetic, may be 2H or 3R or a mixture of both. In nature, 2H-MoS₂ is dominant (Frondel & Wickman, 1970). Further polytypic structures in molybdenite, other than 2H or 3R, have been derived analytically (Zvyagin & Soboleva, 1967) but have not yet been observed experimentally.

Recently, calculations to determine the mean-square displacements (MSD) of 2H-MoS₂ have been made by Feldman & Boyer (1981). In the present paper, we compare these calculations with experimental findings and refine the structural z parameter which is half the height of the MoS₆ prism studied earlier by Dickinson & Pauling (1923) and Takéuchi & Nowacki (1964).

II. Theory

The structures of the 2*H* and 3*R* polytypes of MoS₂ are easily visualized by starting from the unit structure of an MoS₂ plane sandwich (Zvyagin & Soboleva, 1967). This unit structure is shown in Fig. 1, with the S atoms at $(0, 0, \pm z)$, the Mo atom at $(\frac{2}{3}, \frac{1}{3}, 0)$ and with the nomenclature for the different lattice sites *A*, *B* and *C*. We obtain the 2*H* polytype by stacking this unit structure along the *c* axis (leaving a van der Waals gap © 1983 International Union of Crystallography

^{*} Present address: HASYLAB, DESY, Notkestrasse 85, 2 Hamburg 52, Federal Republic of Germany.

[†] Present address: Department of Computer Science, University of Houston, Houston, Texas, USA.



Fig. 1. Projection on the (00.1) plane of the MoS_2 unit repeat structure; O S atoms, O Mo atoms, × lattice origins: A, B, C. The unshaded triangle represents the antiparallel orientation to the MoS_6 prism (shaded).

in between) with a position sequence AB'AB'... for the S atoms, where the prime refers to the antiparallel orientation of the MoS₆ prism (shaded triangle of Fig. 1). If the stacking sequence is ABCABC..., *i.e.* the MoS₆ prisms are oriented parallel, we obtain the 3R polytype. Obviously, the unit cell for 3R-MoS₂ will be larger by a factor of about 1.5 along the c axis when compared with 2H-MoS₂. This is observed experimentally together with a nearly unmodified a parameter [see Murray & Evans (1979) for a recent review on 2H-MoS₂ and Takéuchi & Nowacki (1964) for 3R-MoS₂].

The 2H and 3R polytypes are experimentally distinguished from one another in a straightforward fashion. In Table 1, the structure factors are given for the sets of Bragg reflections that we later employ. The f_x^t values give the scattering amplitude of atom x with the inclusion of the thermal Debye-Waller factor (DWF). For the 3R polytype the same DWF has been tacitly assumed for both S sites. In Table 1, we see that the factor of 2 for 2H-MoS₂ and 3 for 3R-MoS₂ may be taken out, which simply reflects the number of MoS, units per unit cell. It is then clear that the 00.1 reflections may not be used to differentiate between the 2H and 3R polytypes. h0.0 reflections, however, allow us to decide whether 2H is present or not: the reflections $3h' \pm 1$ 0.0 are only allowed for 2H-MoS₂. Finally, the presence of $3R-MoS_2$ can be determined from the h0.l reflections. Although all the 2θ values for low-l reflections are close to one another $(c \ge a)$, the

Table 1. Structure factors |F| for the 2H and 3R polytypes of MoS₂

In comparing |F| with those of Zvyagin & Soboleva (1967), care must be taken because they based their evaluation of F on an expression that neglects dispersion effects in the atomic scattering factors.

hk.l	$ F (2H-MoS_2)$	$ F (3R-MoS_2)$
h0.1	$2!(f'_{Mo} + 2f'_{S} \cos 2\pi lz) \cdot \cos \pi h/3!$ l = 2l'	$3 f'_{Mo} \exp(-2/3\pi hi) + 2f'_{S} \cos 2\pi lz$ h - 3h' and l = 3l'
	$2!(f_{Mo}^{t} - 2f_{S}^{t} \cos 2\pi lz) \cdot \sin \pi h/3!$	h = 3h' + 1 and $l = 3l' + 1h = 3h' - 1$ and $l = 3l' - 1$

0

otherwise

differences separating the ω values of the h0.0 and h0.1 reflections, stemming from the different c parameters of the unit cell for 3*R*- and 2*H*-MoS₂, still permit a clear distinction to be made.

The thermal DWF, and thus the MSD $\langle u_x^2 \rangle = B_x/8\pi^2$ of atom x, has so far been included in the scattering amplitudes without any specification. To account for the two atom types involved and the layered structure of the compound, four thermal DWF's are sufficient to describe MoS₂: B_{Mo}^{\perp} , B_s^{\perp} , B_{Mo}^{\parallel} and B_s^{\parallel} . Here, \perp refers to the displacements along the c direction and \parallel to those within the basal plane as in James (1958). Thus, the f_x^t values for h0.l reflections can be written as

$$f_x^t = f_x \exp\left[-(B_x^{\perp} \cos^2 \varphi + B_x^{\parallel} \sin^2 \varphi) S^2\right],$$

where $S^2 = (\sin \theta / \lambda)^2$ and φ is the angle between the *c* axis and the direction of the scattering vector. Separately, B_x^{\perp} and B_x^{\parallel} can be found from 00.*l* and *h*0.0 reflections, respectively.

III. Experimental details

Two synthetic MoS₂ single-crystal flakes, one of the 2H, the other of the 3R polytype, of irregular lateral form were used. They had been prepared by Dr A. Wold for the Exxon Corporate Science Research Laboratories, Linden, New Jersey, by chemical vapor deposition and were kindly given to us by Dr R. R. Chianelli. Both samples had a [00.1] surface normal in common. Their thicknesses varied from d = 120 to 200 µm as determined from X-ray absorption measurements (using $\mu_l = 74 \text{ cm}^{-1}$). For the 00.*l* reflections the ω scans, or rocking curves, had a FWHM of ~0.15°, whereas for the basal-plane reflections, a larger FWHM of $\sim 0.30^{\circ}$ was found. Their lattice parameters as determined by the Bond method were a = 3.161 and c = 12.295 Å for 2H-MoS₂ and a = 3.163 and c =18.37 Å for 3R-MoS₂, with an error of ~0.05%. For the 3R-MoS₂ crystals, no indication of the 2H polytype was found. The 2H-MoS₂ sample, however, showed a small amount of the 3R polytype (about 5%).

A 2 kW Mo anode sealed X-ray tube (0.4 \times 8 mm) and a Johann (111) Ge monochromator were used, which yielded essentially pure Mo $K\alpha_1$ ($\lambda =$ 0.70926 Å). A small sample spot of ~1 mm height was employed, and an incoming-beam divergence of ~5' in the scattering plane was used. Special care was taken to ensure the selection of a homogeneous sample area. ω scans (conventionally performed by scanning ω at fixed 2θ) with an open counter were performed in a semifocusing set-up to determine the integrated Bragg intensities at room temperature (RT). To separate Bragg intensity and background, the intensities at both ends of the ω scans were linearly interpolated through the Bragg reflection. It is known, however, that the thermal diffuse scattering (TDS) also peaks beneath the Bragg reflection. For a rapid estimation of its contribution to the measured integrated intensities, we employed a method outlined by Jennings (1970). We used average sound velocities in his spherical approximation that were determined from the known elastic data (Feldman & Boyer, 1981).

In the evaluation of the MSD and z parameter from the integrated intensities, the function $\sum w(|F_o| - |F_c|)^2$ was minimized where w is the statistical weight, F_o and F_c are the observed and calculated structure factors, and the residual $R = \sum ||F_o| - |F_c||/\sum |F_o|$. $|F_o|$ was obtained after intensity corrections for background, Lorentz-polarization and absorption. $|F_c|$ was calculated from the scattering factors for Mo and S of Doyle & Turner (1968) and Cromer & Liberman (1970), reflecting that the bonding within an MoS₂ layer is predominantly covalent. [The fractional ionic character of the bonding is discussed by Gamble (1974).]

IV. Results

In the following, our measurements at h0.l reflections will be presented in two groups: those at 00.l reflections and those at h0.0, because they exclusively depend on B_x^{\perp} and B_x^{\parallel} respectively. In the latter group we also included the h0.l reflections with low l on account of their weak dependence on B_x^{\perp} .

1. 00.1 reflections

Reflections ranging from 00.6 to 00.34, the highest reflection order attainable with Mo $K\alpha_1$, were measured with 2H-MoS₂ and 00.9 to 00.48 with 3R-MoS₂ (Table 2). For 2H-MoS₂ it was checked that no 00.*l* reflections with odd *l* were present as these had been previously seen in measurements with crystals of natural 2H-MoS₂ (Huang, 1981) and were probably due to its contamination with metallic impurities and water.

In the fitting procedure, B_{Mo}^{\perp} , B_{S}^{\perp} , and z were taken as free parameters and their final values are given in Table 3. For the final data fit presented in Table 2, the two lowest reflections were not included, because their ω scans indicated surface-related problems with lower ω values. The R value was 1.9% for the 2H-MoS₂ data and 2.6% for the 3R-MoS₂ data. Three conclusions can be drawn from Table 3. First, both polytypes show good agreement between all three fitting parameters B_{Mo}^{\perp} , B_{S}^{\perp} and z. Second, the B^{\perp} value of the Mo atoms is distinctly larger than that of the S atoms. This is somewhat unexpected, for the Mo atoms are heavier than the S atoms. Third, B_{Mo}^{\perp} and B_{S}^{\perp} increase only slightly when we consider the TDS corrections according to Jennings (1970) without changing the first two conclusions.

Table 2. Observed $|F_o|$ and calculated $|F_c|$ structure factors for 2H- and 3R-MoS₂

2H-MoS ₂			$3R-MoS_2$		
hk.l	$ F_o $	$ F_c $	hk.l	$ F_o $	$ F_c $
00.6*	46.74	62.85	00.9*	80.55	93.74
00.8*	67.18	84.13	00.12*	101.6	126.1
00.10	39.36	39-21	00.15	57.37	59-42
00.12	12.36	13.59	00.18	19.22	20.54
00.14	38.42	38.05	00.21	58-63	56-50
00.16	48-41	48.05	00.24	71.97	72-14
00.18	20.54	20.08	00.27	31.12	30.94
00.20	7.79	8.48	00.30	12.71	13.02
00.22	21.93	22.06	00.33	34.22	32.70
00.24	24.70	24.58	00.36	37-02	37-0
00.26	10.12	9.39	00.39	15.04	14.7
00.28	5.00	4.88	00.42	7.60	7.6
00.30	11.37	11.22	00.45	17.66	16-6
00.32	11.17	11.41	00.48	16.19	17-29
00.34	3.80	3.95			
10.0	58.09	56-49			
20.0	42.73	40.20			
30.0	56-42	61.18			
40.0	23.47	23.06			
50.0	16.56	16.79			
60.0	24.62	23.98			
70.0	8.42	8.44			
$10.\pm 1$	29.90	30.11			
$20.\pm 1$	23.78	23.26			
$40.\pm1$	13.60	13.83			
$50.\pm1$	10.63	10-88			
$70.\pm1$	6.19	6.19			

* Reflections not included in the final refinement due to surface effects at low ω .

Table 3. Positional and thermal parameters of 2H- and 3R-MoS₂

\perp refers to vibrations normal to the basal plane (along the c axis). \parallel refers to in-plane vibrations.

	$egin{aligned} B^{\perp}_{ m Mo}({ m \AA}^2)\ (\langle u^{\perp\ 2}_{ m Mo} angle imes 10^3{ m \AA}^{-2}) \end{aligned}$	$egin{aligned} B_{ m S}^{\perp}({ m \AA}^2)\ (\langle u_{ m S}^{\perp 2} angle imes 10^3~{ m \AA}^{-2}) \end{aligned}$	z
2H-MoS ₂	0.61 (7.7)	0.39 (4.9)	0.127,
	0.62*	0.40*	0.1275
3R-MoS	0.59 (7.5)	0.40 (5.1)	0·127, × ≩
2	0.61*	0.41*	$0.127_{4} \times \frac{2}{3}$
	$\frac{B_{\rm Mo}^{\scriptscriptstyle \parallel}({\rm \AA}^2)}{(\langle u_{\rm Mo}^{\scriptscriptstyle \parallel}^{\scriptscriptstyle \perp}\rangle\times 10^3{\rm \AA}^{-2})}$	$\begin{array}{c} B_{\rm S}^{\scriptscriptstyle \parallel}({\rm \AA}^2)\\ (\langle u_{\rm S}^{\scriptscriptstyle \parallel 2}\rangle \times 10^3{\rm \AA}^{-2}) \end{array}$	
2H-MoS ₂	0.38 (4.8) 0.40*	0·30 (3·8) 0·32*	

* Data were corrected for TDS according to Jennings (1970). The final values of $\langle u^2 \rangle$ are not altered within the allowed error.

2. h0.0 and $h0.\pm 1$ reflections

Reflections that mainly yield the MSD within the basal plane were measured with both crystals. In the case of 3R-MoS₂, however, a large R value was obtained, most probably arising from stacking faults in the crystal used. For this reason we will only present the results for 2H-MoS₂.

All h0.0 and h0. ± 1 reflections obtainable with Mo $K\alpha_1$ were measured (Table 2). In our analysis the

 $h0.\pm 1$ reflections were averaged after the absorption corrections. B_{Mo}^{\parallel} and B_{S}^{\parallel} were used as free parameters with the above results for B_{Mo}^{\perp} , B_{S}^{\perp} and z. The fit gave R = 3.6%. The result for B_{x}^{\parallel} is smaller than B_{x}^{\perp} for the Mo and S atoms. In addition, their B values differ less within the basal plane than along the c axis. This would seem to reflect the closer packing within these planes.

V. Discussion

Our determination of the MSD for both 2H- and 3R-MoS₂ shows the following characteristics:

(i) the MSD along the c axis for the Mo atoms is larger than that for the S atoms and does not depend on the polytype;

(ii) both atoms have smaller and more comparable MSD's within the basal plane.

These results may be compared with calculations for the MSD of 2H-MoS₂ by Feldman & Boyer (1981). At RT they calculated $\hat{B}_{Mo}^{\perp} \simeq 0.47, B_{s}^{\perp} \simeq 0.51, B_{Mo}^{\parallel} \simeq 0.29$ and $B_{s}^{\parallel} \simeq 0.44 \text{ Å}^{2}$. Thus the measured and the calculated data fall within the same range and the anisotropy parameters $A = B_x^{\perp}/B_x^{\parallel}$ for the Mo atoms (1.6 measured, 1.60 calculated) and for the S atoms (1.3 measured, 1.15 calculated) also agree reasonably well. However, the calculated ratio between the two atom types is not in agreement with the experimental findings; in particular, the calculated MSD along the caxis is smaller for the Mo atoms than for the S atoms. Analogous experimental findings have also been seen with other layered dichalcogenides, independent of the details of their structures: TiS₂, which has a CdI₂ structure (Chianelli, Scanlon & Thompson, 1975), and 3R-NbS₂ (Morosin, 1974) both show a larger MSD along the c axis for the metal atoms than for the S atoms, but by a smaller amount of 10–15%.

The reason for the larger c-axis MSD for Mo over S is not obvious. It is perhaps related to the tendency in the trichalcogenides, such as $NbSe_3$ (Fleming, Moncton & McWhan, 1978), towards dimerization of the metal

atoms which may be connected to a (general) longitudinal softness along c of the metal atom in a sixfold chalcogen-coordinated site.

Finally, the z parameter may be compared with the literature values. Our results of 0.127_5 for 2H-MoS₂ and $\frac{2}{3} \times 0.127_3$ for 3R-MoS₂ with synthetic crystals are quite close to $z = 0.129 \pm 0.004$ for 2H-MoS₂ reported by Dickinson & Pauling (1923) and $z = \frac{2}{3} \times 0.128_5$ for 3R-MoS₂ found by Takéuchi & Nowacki (1964) with natural samples.

This research was supported by the USARO through a joint grant to Michigan State University and the University of Houston. One of us (BS) thanks the Alexander von Humboldt Foundation for his Feodor Lynen Fellowship and the USDOE for partial support under contract DE-AS05-76ER05111. We thank R. R. Chianelli for the crystals, prepared for him by A. Wold of Brown University.

References

- CHIANELLI, R. R., SCANLON, J. C. & THOMPSON, A. H. (1975). Mater. Res. Bull. 10, 1379-1382.
- CROMER, D. T. & LIBERMAN, D. (1970). J. Chem. Phys. 53, 1891–1898.
- DICKINSON, R. G. & PAULING, L. (1923). J. Am. Chem. Soc. 45, 1466-1471.
- DOYLE, P. A. & TURNER, P. S. (1968). Acta Cryst. A24, 390-397.
- FELDMAN, J. L. & BOYER, L. L. (1981). Solid State Commun. 37, 879–881.
- FLEMING, R. M., MONCTON, D. E. & MCWHAN, D. B. (1978). *Phys. Rev. B*, 18, 5560–5563.
- FRONDEL, J. W. & WICKMAN, F. E. (1970). Am. Mineral. 55, 1857-1874.
- GAMBLE, F. R. (1974). J. Solid State Chem. 9, 358-367.
- HUANG, J. J. (1981). Thesis for the MS degree, Univ. of Houston
- (unpublished). JAMES, R. W. (1958). The Optical Principles of the Diffraction of X-rays. London: Bell.
- JENNINGS, L. D. (1970). Acta Cryst. A 26, 613-622.
- MOROSIN, B. (1974). Acta Cryst. B30, 551-552.
- MURRAY, R. & EVANS, B. L. (1979). J. Appl. Cryst. 12, 312-315.
- TAKÉUCHI, Y. & NOWACKI, W. (1964). Schweiz. Mineral. Petrogr. Mitt. 44, 105-120.
- WILSON, J. A. & YOFFE, A. D. (1969). Adv. Phys. 18, 193-335.
- ZVYAGIN, B. B. & SOBOLEVA, S. V. (1967). Sov. Phys. Crystallogr. 12, 46-52.